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Effect of ALD Processes on Physical and Electrical Properties of HfO₂ Dielectrics for the Surface Passivation of a CMOS Image Sensor Application

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ABSTRACT The surface passivation of a CMOS image sensor (CIS) is highly beneficial for the overall improvement of a device performance. We employed the thermal atomic layer deposition (T-ALD) and plasma enhanced (PE-ALD) techniques for the deposition of 20 nm HfO₂ as well as stacked with 3 and 5 nm Al₂O₃ thin films. The HfO₂/Si and Al₂O₃/HfO₂/Si metal-oxide-semiconductor structures were used to analyze the fixed charge density (Q_f) and interface trap density (D_{it}). The as-synthesized samples show high D_{it} and Q_f values (10¹² cm⁻²eV⁻¹) and a minority carrier lifetime of 15-300 μs. The finite-difference time-domain simulation of high-k dielectrics confirmed that the Al₂O₃ (top)/HfO₂ stacked structures expected higher quantum efficiency for CIS application. The effect of vacuum annealing (VA) and forming gas annealing (FGA) treatments succeeded with the decomposition of the D_{it} and increase in carrier lifetime. The H₂ ambient FGA samples showed a remarkable decrease in the D_{it} values. To improve the overall performance of the device after passivation, we employed an Al₂O₃/HfO₂ bilayer structure, which showed a low D_{it} of 10¹¹ cm⁻²eV⁻¹ and a minority carrier lifetime of ~3,700 μs after 400 °C and 30 min FGA. We believe that this surface passivation strategy will pave way for future CIS technology regarding the development of lower defective surface and superior performance.

INDEX TERMS Atomic layer deposition, carrier lifetime, CMOS image sensor, surface passivation, forming gas, HfO₂, interface trap density.

I. INTRODUCTION

The rapid progress in CMOS circuit technology is expecting a dominant market extension regarding novel electronic devices [1]. CMOS technology plays an important role in the evolution of image sensors as modern low-cost sensors require high resolution, high speed, and low power consumption [2]. However, the performance of the CMOS image sensor (CIS) is highly affected by defects (Si-dangling bonds) and impurities that are linked with Si interfaces and surfaces. Therefore, it is important to passivate the Si surface to achieve better device performance.

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The defects at the Si/dielectric interface introduce different energy levels into the Si bandgap, which are generally termed as interface trap density (D_{it}). Therefore, excellent chemical surface passivation is required to suppress the defects at the Si/dielectric interface, which can reduce the D [3]. On the other hand, the fixed charge (Q_f) is the charge accumulation within a few nm of the Si/dielectric interface. The Q_f is the resultant of the surface properties and annealing conditions. The presence of charges in the dielectric film established an electric field at the interface, which can either attract or repel the carriers from the surface [4]. Thus, attracting or repelling one type of carrier from the surface limits the charge recombination rate [5]. Also, the negative charges for the p-type Si and the positive charges for the n-type Si

can improve the quality of the field-effect passivation in the entire injection level range. The quality of passivation by the dielectric is commonly examined by the minority carrier lifetime [6]. The minority carrier lifetime depends on the surface recombination and the dark current of the thermally generated electrons at the interface. This results in the excitation of electrons from the valence band to the conduction band level [5].

Up to now, a variety of high-k dielectric materials, such as Al₂O₃, HfO₂, Ta₂O₅, and TiO₂ are of beneficial interest for the surface passivation [7]–[9]. Among these high-k materials, HfO₂ and Al₂O₃ dielectrics are of receiving progressive attention due to their outstanding defect decomposition properties in regard to the CIS applications [10]. Interestingly, the bilayer or laminated structures of the dielectric materials, such as Al₂O₃, HfO₂, SiO₂, and TiO₂ are widely adopted due to their overall surface passivation benefits and improved device performance [7], [8]. However, the high quality of chemical as well as field effect passivation is introduced by Al₂O₃ and HfO₂ which effectively passivates the defects. This is because of the improvement in the interfacial layer due to the bilayer structure of Al₂O₃ and HfO₂. Owing to there excellent interfacial characteristics, these Al₂O₃/HfO₂ bilayer structures are getting more and more attention for the improvement of passivation quality. The synthesis of the dielectric thin films has been implemented by a variety of techniques [11]. Atomic layer deposition (ALD) is a state-of-the-art and crucial technique among the various techniques to synthesize different dielectric materials. The ALD process represents the key features of the as-synthesized films, which include a highly effective uniform deposition, low-temperature process, and layer-by-layer growth. Also, it is most favorable as compared to other techniques due to a self-limited reaction mechanism that is based on the chemisorption of two or more precursors [12], [14].

In this research work, we have investigated the different process parameters as well as the post-metal annealing conditions that realized the decomposition of the D_{it} and increased Q_f for a single HfO₂ structure. However, to increase the carrier lifetime and improve the interface, we have employed the bilayer Al₂O₃/HfO₂ laminate structures. The results show that the H₂ ambient forming gas annealing (FGA) remarkably decreases the D_{it} values than vacuum annealed (VA) samples. The Al₂O₃/HfO₂ bilayer structure may shows better field-effect and chemical passivation effect [18] in an integration of CIS process by minimizing the formation of interfacial layer (Si/dielectric) and improve characteristics of the dielectric properties. Hence, after FGA of Al₂O₃/HfO₂ bilayer structure, the interfacial layer gets improved which results in reduction of D_{it} values and recombination. Also, combining effect of these high-k materials effectively enhances the passivation quality. We believe that this surface passivation strategy will pave a way for future CIS technology regarding the development of lower defective surface and superior performance.

II. EXPERIMENTAL DETAILS

The HfO₂ ALD films were deposited using a tetrakis (ethylmethylaminohafnium) (TEMAHf, Hf[N(C₂H₅)(CH₃)₄]) precursor at a temperature range between 100 °C to 400 °C. The p-type Si ((100), 18 Ω-cm) substrates were cleaned using an ammonia peroxide mixture (NH₃:H₂O₂:de-ionized (DI) water with a ratio of 1:1:5 at 70 °C for 10 min) and diluted hydrofluoric acid (50:1, DI H₂O) at room temperature for 1 min before the deposition. The HfO₂ films were then deposited using ALD reactor (Plus 200 system, Quoros). The TEMAHf precursor was supplied with O₂ plasma at a 200 sccm flow rate and a power of 100 W, for the sequential surface reactions in the plasma-enhanced ALD (PE-ALD) at a chamber pressure of 0.4 mTorr. Also, H₂O precursor was used in thermal ALD (T-ALD) for the 20-nm-target film thickness instead of the O₂ plasma for PE-ALD deposited films. For comparison of the different HfO₂ films synthesized using different ALD processes, the T-ALD HfO₂ and PE-ALD HfO₂ samples were prepared at different deposition temperatures (250 °C and 280 °C) on the pre-cleaned Si wafer. The post-deposition annealing was performed in a tube furnace at vacuum and forming gas (N₂:H₂ = 95 %:5 %) conditions for all the HfO₂ samples. The annealing temperature was kept constant at 400 °C for 30 min.

To fabricate the standard Al/HfO₂/Si and Al/Al₂O₃/HfO₂/Si MOS capacitor (MOSCAP) like structures, a 300 nm Al top electrode was deposited using thermal evaporation. The custom-made shadow mask with a 150 × 150 μm² dimension was used for the dot structure as a gate electrode for all the structures. The capacitance-voltage (C-V) and the current density-voltage (J-V) measurements of the prepared structures were then conducted using the Agilent B1500A parameter analyzer. All the measurements were traced at room temperature with a high frequency (1 MHz) under dark conditions. The dual C-V curves of 10 devices in the forward and backward directions were collected from the accumulation to the inversion region properties of the MOSCAPs. The J-V characteristics were then conducted to investigate the effect of the different passivation layers on the electrical properties of the prepared MOSCAPs.

The D_{it} and Q_f values were estimated using C-V curves from Eqs. 1 and 2, respectively [15].

$$Q_f = \frac{(\phi_{ms} - V_{FB})C_{ox}}{qA}, \quad (1)$$

where ϕ_{ms} is defined by the difference between the metal and the semiconductor work functions, V_{FB} is the flat band voltage, C_{ox} is the oxide capacitance, q is the electron charge, and A is the electrode area.

$$D_{it} = \frac{2\omega C_{ox}^2 G_{max}}{qA(G_{max}^2 + \omega^2 (C_{ox} - C_m C_{max})^2)}, \quad (2)$$

where G_{max} is the maximum conductance, and C_m is the measured capacitance at the frequency ω .

The thickness and refractive index (n) of the HfO₂ films were determined using spectroscopic ellipsometry (alpha SE,

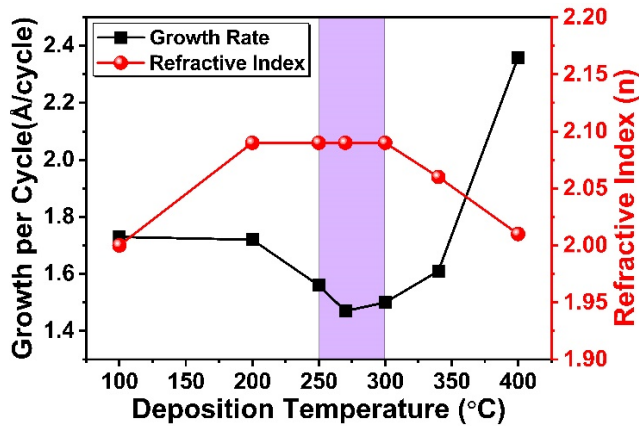


FIGURE 1. Growth per cycle (GPC) (left axis) and the refractive index (n) at 632.8 nm (right axis) of the HfO₂ films on the Si wafer at substrate temperatures between 100 °C and 400 °C.

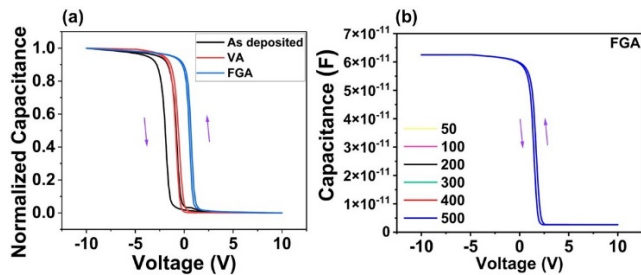


FIGURE 2. The C-V curves of the HfO₂ samples synthesized at 250 °C using plasma enhanced-ALD technique under different annealing conditions (a), and multiple C-V cycles of FGA sample (b).

J. A Woollam co., Ltd). The minority carrier lifetime (injection level-dependent) of the Si-wafers was measured using the quasi-steady-state photoconductance (QSS-PC) method with the lifetime tester (WCT-120, Sinton Instruments) [16]. The minority carrier lifetime was extracted at an intermediate carrier level injection of $1 \times 10^{15} \text{ cm}^{-3}$.

III. RESULTS AND DISCUSSION

In this study, the ALD synthesis of the HfO₂ thin films was conducted at a wide temperature range from 100 °C to 400 °C using a TEMAHf precursor to examine the thorough deposition variation. Fig. 1 illustrates the growth per cycle (GPC) and n of the HfO₂ thin films with respect to the deposition temperature. The GPC vs the deposition temperature curve (black) in Fig. 1 indicates that the GPC decreases at 200 °C and reaches a stable value of about 1.5 Å/cycle in the range of 250 °C to 300 °C. After that, the GPC increased in the temperature range from 300 °C to 400 °C. Thoroughly, the initial temperature range (100-200 °C) possesses minimal activation energy for the formation and growth of the HfO₂ thin films [17]. On the other hand, temperature above 300 °C showed a CVD-type growth performance, which showed a high GPC value [18]. This implies that the temperature range of 250-300 °C is a crucial temperature range for the optimal process window. Additionally, the ellipsometry

measurements showed a slanted increase of the n from 100 °C to 200 °C. However, with a further increase in temperature, the n values acquire a stable range from 250 °C to 300 °C, which again decreased for both ALD processes. Thus, the overall results imply that the ALD process window is optimum in the range of 250-300 °C. Fig. 2(a) shows the C-V characteristics of the as-deposited PE-ALD HfO₂ samples at 250 °C under different annealing conditions. The arrows indicate the voltage sweep direction in the C-V curves. The obtained C-V curves display hysteresis in the as-prepared HfO₂ films. The hysteresis in the C-V curves is related to the existence of a certain amount of interface defects at HfO₂/Si interface. [19], [20]. However, after VA and FGA treatments there is a remarkable reduction in the C-V hysteresis due to a reduction in interface defects [19], [21]. Further, after VA and FGA, the V_{FB} showed a negative value, which reflects the existence of trapped positive oxide charges in the HfO₂ passivation layer. The T-ALD films also exhibited similar characteristics as PE-ALD grown films.

In addition to V_{FB}, the permittivity (K) and other parameters of the as-grown films are summarized in Table 1. The EOT and the K of the HfO₂ films were extracted using the relation provided below Eqs. 3 and 4. [22]:

$$EOT = \frac{K_{SiO_2} \times \epsilon_0}{C_{max}/A}, \quad (3)$$

$$K = \frac{K_{SiO_2} \times t_{hk}}{EOT - t_{SiO_2}}, \quad (4)$$

where K_{SiO_2} is the permittivity of SiO₂, A is the area of the Al electrode, C_{max} is the accumulation capacitance, and t_{hk} is the thickness of the HfO₂ films. The annealing effect on the HfO₂ films prepared at 250 °C in PE-ALD showed an increase in the EOT and a reduction in K, as illustrated in Table 1. For the films prepared at 280 °C (T-ALD), 250 °C (PE-ALD), and 280 °C (PE-ALD), the same trend was observed. It can be concluded that the HfO₂ passivation layer can be attributed to the formation of the interface layer after the annealing [23], [24].

The J-V characteristics of the 280 °C PE-ALD grown films with VA and FGA conditions were measured after applying a negative bias. We observed a very low current density of $4.5 \times 10^{-7} \text{ A/cm}^2$ at -1 V after the annealing process at 400 °C. Due to the existence of a small number of dangling bonds and an interface charge trapping, the leakage current gets reduced after FGA [22], [25].

Fig. 3 shows the role of VA and FGA on the D_{it} values of the 20-nm-thick HfO₂ thin films deposited using T-ALD and PE-ALD growth techniques. For both synthesis techniques, the as-prepared films possessed high D_{it} values ($1.2 \sim 1.5 \times 10^{12} \text{ cm}^{-1} \text{ eV}^{-1}$) before the annealing process. However, T-ALD showed lower D_{it} values as compared to the PE-ALD technique. The Si-dangling bonds and the inferior HfO₂ and Si interface seemed to cause the high D_{it} values for the as-prepared samples [15], [16]. Interestingly, in both ALD processes, the as-prepared HfO₂ films show a remarkable reduction in the D_{it} values ($6 \sim 11 \times 10^{11} \text{ cm}^{-1} \text{ eV}^{-1}$), which

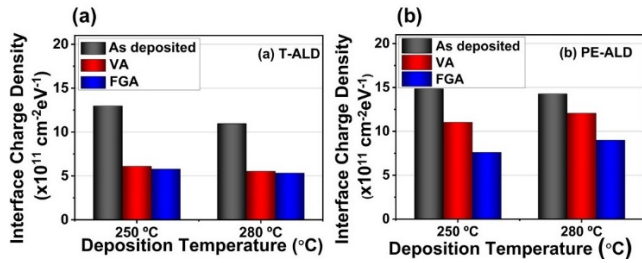


FIGURE 3. Interface trap density (D_{it}) for 20-nm-thick HfO₂ films after annealing at 400 °C for 30 min. (a) thermal-ALD grown HfO₂ films and (b) plasma enhanced-ALD grown HfO₂ films.

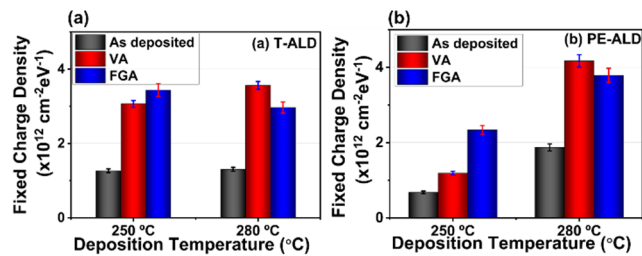


FIGURE 4. Fixed charge density (Q_f) for 20-nm-thick HfO₂ films after annealing at 400 °C for 30 min. (a) thermal-ALD grown HfO₂ films and (b) plasma enhanced-ALD grown HfO₂ films.

visualizes the significant role of the annealing parameters. The device possessed minimal D_{it} values after the FGA process. The obtained results indicate that the H₂ annealing at 400 °C improves the quality of the interface by hydrogenation, which is advantageous for strong chemical passivation of the Si-dangling bonds as opposed to the VA [26], [27]. This visualization depicts the dominant role of the FGA over the VA samples.

In Fig. 4, the Q_f values were measured as a function of the different HfO₂ synthesis temperature to study the field-effect passivation [27]. For this analysis we have measured 10 devices. The error bars in Fig. 4 represents that there is not much significant deviation in the analyzed samples. With the HfO₂, the equivalent series resistance, and other models predict that the positive charges near or at the interface of the metal/semiconductor were present due to oxygen vacancies [8], [27]. The as-prepared T-ALD films demonstrated high Q_f values for low substrate temperatures. However, for PE-ALD grown films, Q_f values get increased at high substrate temperature samples. The FGA samples showed higher Q_f values for T-ALD and PE-ALD methods than as-deposited HfO₂ films. The increased Q_f values after VA and FGA conditions may be caused by interfacial layer formation, revealing the better field-effect passivation than as-synthesized HfO₂ films [28].

After the FGA, the D_{it} values decreased, and the Q_f values increased which implies the collective effect of the chemical as well as the field-passivation. This is an ideal phenomenon to possess strong surface passivation [29], [30].

Finite-difference time-domain (FDTD) is used to calculate the reflectance, transmission, interference, diffraction, and

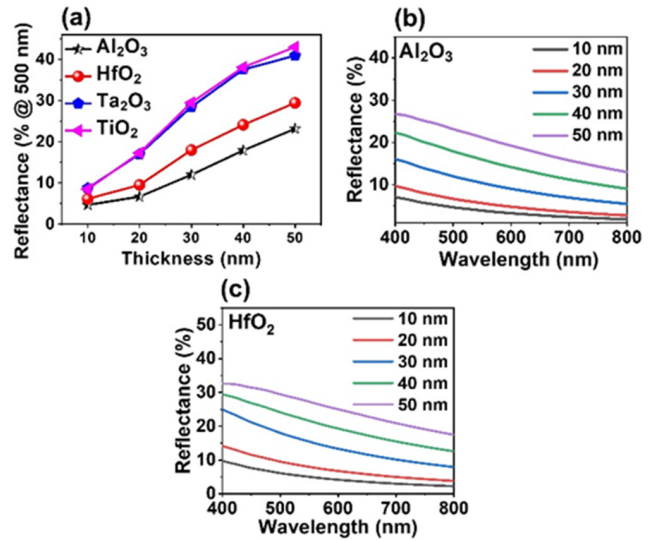


FIGURE 5. Finite-difference time-domain (FDTD) simulation results: (a) reflectance at 500 nm incident wavelength as a function of thickness for each high-k dielectric, (b) reflectance as a function of wavelength for the Al₂O₃ films, and (c) reflectance as a function of wavelength for the HfO₂ films.

absorption of structured or unstructured materials, as well as to study the response to different polarizations and wavelengths of light and demonstrate how light travels through the materials in real-time [31]. Fig. 5 represents FDTD simulation to evaluate the reflectance measurements of the single-layer and bilayer structures. In Fig. 5(a), the reflectance (%) at a 500 nm wavelength for different thickness of Al₂O₃, HfO₂, Ta₂O₅, and TiO₂ films on the Si substrates is shown. The graphical representation of reflectance at 500 nm was found to be linearly proportional to the thickness in the all-dielectric materials in the simulation. However, the lower reflectance values of the Al₂O₃ and HfO₂ dielectrics result in higher quantum efficiency (QE) values than Ta₂O₅ and TiO₂. Nevertheless, the comparison between the reflectance of Al₂O₃ and HfO₂ reveals that Al₂O₃ results in better QE than HfO₂ films (Fig. 5(b) and Fig. 5(c)). Furthermore, the Al₂O₃/HfO₂ stack structure results in high QE, and it is helpful for the CIS applications with its low reflectance. The reflectance of the Al₂O₃ and HfO₂ dielectrics results in higher quantum efficiency (QE) values than Ta₂O₅ and TiO₂. Nevertheless, the comparison between the reflectance of Al₂O₃ and HfO₂ reveals that Al₂O₃ results in better QE than HfO₂ films (Fig. 5(b) and Fig. 5(c)). Furthermore, the Al₂O₃/HfO₂ stack structure results in high QE, and it is helpful for the CIS applications with its low reflectance.

The generation and recombination of the carriers at the interface due to defects might result in the emergence of a dark current which can demean the QE of the CIS [32]. On the other hand, the reduction of carrier recombination is the key feature in fabricating Si-based high-performance cutting-edge electronic devices [33]. Figs. 6(a) and 6(b) demonstrate the effective carrier lifetime of the as-prepared and the FGA HfO₂ devices that were prepared by T-ALD and PE-ALD.

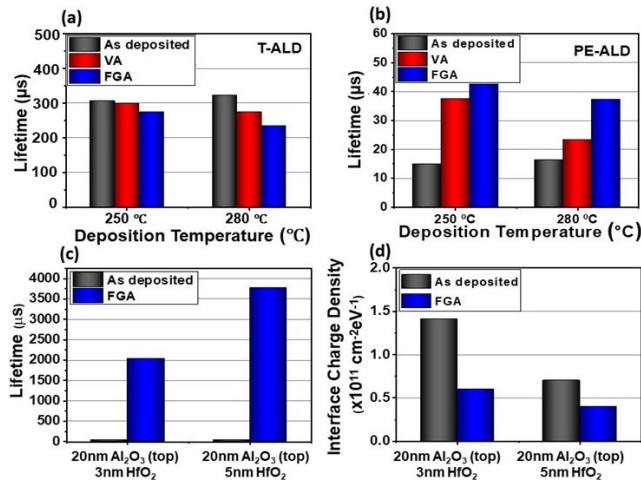


FIGURE 6. Effective carrier lifetime before and after annealing (400 °C for 30 min) of (a) thermal-ALD HfO₂, (b) plasma enhanced-ALD HfO₂, (c) plasma enhanced-ALD 3 nm and 5 nm of the Al₂O₃(top)/HfO₂ stacked structure films, and (d) the interface trap density (D_{it}) measurements of the Plasma enhanced-ALD 3 nm and 5 nm Al₂O₃ (top)/HfO₂ stacked structure films.

In Fig 6(a), T-ALD grown HfO₂ films show similar values under all conditions. However, the as-prepared samples exhibit lower carrier lifetime values as compared to the FGA samples, which is shown in Fig. 6(b). The low lifetime values of the as-prepared samples occur due to the recombination centers induced by the interfacial defects [34]. However, the carrier lifetime value of the single-layer HfO₂films gets improved after the FGA (300 μs). The minimized charge carrier recombination is attributed to the enhanced lifetime after the FGA processing conditions. For further improvement in the carrier lifetime, the bilayer Al₂O₃/HfO₂stacked structure was utilized as predicted by the FDTD simulation (Fig 5). Fig. 6(c) depicts the carrier lifetime measurements of the Al₂O₃/HfO₂bilayer structures, deposited by PE-ALD with 3-nm and 5-nm thick HfO₂ and 20-nm-thick of Al₂O₃. As predicted, bilayer stacked structures of Al₂O₃/HfO₂ showed a higher carrier lifetime than the single-layer HfO₂samples. The high quality of the chemical passivation can reduce the defects, which can increase the carrier lifetime. Later, both the Al₂O₃/HfO₂bilayer structures were subjected to the FGA processing. The Al₂O₃/3-nm HfO₂ structure showed a carrier lifetime of 2,100 μs. However, the Al₂O₃/5-nm HfO₂structure showed a maximum carrier lifetime of 3,750 μs owing to the better interface caused by the 5-nm-thick HfO₂ with Si. The record performance of the bilayer structures shows that the interface traps or the Si-dangling bonds were banished and effectively passivated by the hydrogenation [34], [35]. Further, Fig. 6(d) shows that the Al₂O₃/HfO₂bilayer structure shows lower D_{it} values (1.0 × 10¹¹ cm⁻².eV⁻¹) than the single-layer HfO₂ structures (6.0 × 10¹¹ cm⁻².eV⁻¹) after the FGA processing conditions. Thus, the enhancement in the carrier lifetime due to hydrogenation after the annealing affects the chemical passivation of Al₂O₃/HfO₂, resulting in the record performance.

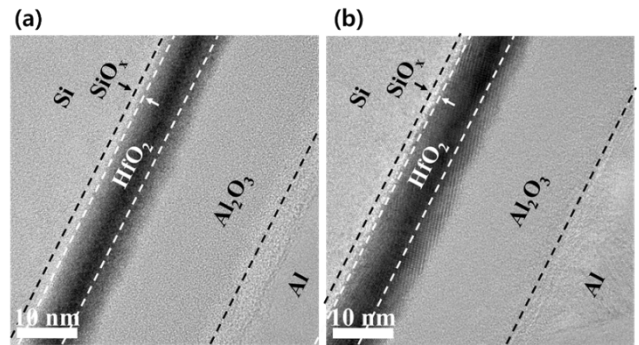


FIGURE 7. Transmission electron microscopic images of the Si/5-nm HfO₂/20-nm Al₂O₃/Al structure (a) before forming gas anneal (FGA) and (b) after FGA.

TABLE 1. Parameters for the HfO₂ and Al₂O₃/HfO₂ passivation layers include flat band voltage (V_{FB}), permittivity (K), equivalent oxide thickness (EOT), and leakage current density (J).

| Deposition | Annealing | V _{FB} (V) | EOT (nm) | K _{ox} | J _g (A/cm ²) |
|--|--------------|---------------------|----------|-----------------|-------------------------------------|
| 250 °C T-ALD | As-deposited | - 1.9 | 5.8 | 20.6 | 5.0 x 10 ⁻⁷ |
| | VA | - 1.4 | 8.5 | 12.0 | 1.7 x 10 ⁻⁷ |
| | FGA | - 1.3 | 9.2 | 10.8 | 1.8 x 10 ⁻⁷ |
| 280 °C T-ALD | As-deposited | - 1.4 | 5.1 | 25.6 | 1.0 x 10 ⁻⁶ |
| | VA | - 1.1 | 6.6 | 17.0 | 3.5 x 10 ⁻⁷ |
| | FGA | - 1.0 | 9.2 | 10.9 | 1.7 x 10 ⁻⁷ |
| 250 °C PE-ALD | As-deposited | - 2.6 | 5.9 | 19.8 | 1.5 x 10 ⁻⁶ |
| | VA | - 1.8 | 6.4 | 17.7 | 1.3 x 10 ⁻⁷ |
| | FGA | - 0.8 | 8.1 | 12.8 | 1.1 x 10 ⁻⁷ |
| 280 °C PE-ALD | As-deposited | - 2.4 | 6.0 | 19.7 | 2.8 x 10 ⁻⁶ |
| | VA | 1.1 | 6.7 | 16.5 | 1.6 x 10 ⁻⁷ |
| | FGA | 1.0 | 6.8 | 16.3 | 4.5 x 10 ⁻⁷ |
| 5nm HfO ₂ / 20nm Al ₂ O ₃ | As-deposited | - 0.4 | 8.6 | 11.7 | - |
| | FGA | 0.4 | 8.8 | 11.5 | - |

To study the interface property of Al₂O₃/5-nm HfO₂ bilayer structures, we used cross-sectional transmission electron microscopy (TEM). In Fig. 7, the TEM micrographs display the different MOS structures with an interface layer between HfO₂/Si. It was observed that the as-deposited Si/5-nm (target thickness) HfO₂/20-nm Al₂O₃(target thickness)/Al samples showed a clear interface between the Si and HfO₂ with a 6.2-nm-thick HfO₂ and a 20.8-nm-thick Al₂O₃. These calculated values from the TEM are corroborated with the expected thickness from the ALD process. The fully amorphous nature of the oxide layers can be observed in the as-deposited ALD structures.

After FGA processing, the thickness of the interface between the HfO₂ and Si showed an improvement, and crystallinity gets increased as shown in Fig. 7(b) [8]. The increased interface thickness is crucial for the reduction of D_{it} values [36]. In addition to this, hydrogenation played a very crucial role in the suppression of the Si dangling bonds, which also accounted for the D_{it} suppression. Thus, the densification of the interface of the bilayer Al₂O₃/HfO₂after FGA will result in decreased EOT and increased K_{ox} as shown in Table 1.

The overall finding of this study revealed that the defects that are associated with the Si surface and its interfaces with oxides can be potentially controlled by chemical and field-effect passivation, which is emerging with CIS technology as well as Si-based electronic devices.

IV. CONCLUSION

In this study, the plasma-enhanced ALD (PE-ALD) and thermal atomic layer deposition (T-ALD) HfO₂ thin films were prepared. The ALD process window was optimum in the temperature range from 250 °C to 300 °C. The high deposition temperature showed higher fixed charge density (Q_f) values in both ALD techniques. In the case of interface trap density (D_{it}), the values for the T-ALD prepared HfO₂ films were lower than the values for the PE-ALD grown films, which indicates that the moderate temperature played a crucial role in the surface passivation for the ALD processes. To improve the overall performance, we studied the Al₂O₃/HfO₂ stack structures. The finite-difference time-domain (FDTD) simulation finding for the Al₂O₃/HfO₂ stack showed higher QE for the CMOS image sensor (CIS) applications. The Al₂O₃/HfO₂ stack layers possessed a minimum D_{it} of $\sim 1.0 \times 10^{11} \text{ cm}^{-2} \text{ eV}^{-1}$. The H₂ ambient forming gas anneal (FGA) (400 °C for 30 min) is advantageous for chemical passivation in CIS applications. The bilayer Al₂O₃/HfO₂ stacked structures showed an increased minority carrier lifetime from $\sim 300 \mu\text{s}$ up to $\sim 3,700 \mu\text{s}$. The analyzed structures can help to increase the carrier lifetime as well as decrease the D_{it} values, which also reveal the importance of the FGA, and can assist with the advancement of the upcoming CIS technology.

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