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Degradation Observations of Protective Coatings: An Improved Version of Coating Impedance Detector

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ABSTRACT This study proposes coating impedance detector 3.0 (CID 3.0), an improved version of our previously developed CID 2.0. The new circuit design in CID 3.0 has lower power consumption because it has fewer components, and it affords better accuracy through the modification of the analog part of CID 2.0 and the use of oversampling. This approach successfully afforded CID 3.0 with higher measurement stability for evaluating a high-performance coating with impedance values exceeding $10^9 \Omega$. Furthermore, CID 3.0 could detect impedance decreases associated with coating delamination when the coating suffers an attack.

INDEX TERMS Coating degradation, corrosion monitor, field-programmable gate array, oversampling, electrochemical impedance spectroscopy.

I. INTRODUCTION

Among many corrosion prevention methods, such as sacrificial anode and impressed current cathodic protection, protective organic coatings are considered the most low-cost, direct, and effective solution for offshore wind turbines. Such coatings, in a manner similar to those on ships and buildings, can provide a physical barrier to protect offshore wind turbines from corrosive environments. However, the coating health gradually deteriorates with time. In other words, coatings cannot provide permanent protection to a substrate, and they must be periodically inspected and maintained to maximize their lifetime and ensure their function. Offshore environments are extremely corrosive owing to the presence of numerous corrosion factors, including chloride ions in seawater and ultraviolet rays in sunlight. The wet-dry cycling conditions at the splash zone are considered another major factor causing the degradation of coatings. Therefore, the health of coatings applied on steel structures should be evaluated periodically to prevent corrosion failures in offshore wind towers.

As an alternative to the well-known traditional testing methods (e.g., the salt spray test [1] and QUV accelerated

weathering test [2]), in recent years, many studies have investigated nondestructive testing (NDT) and strategies for evaluating the properties of a coating-specifically, acoustic emission [3], ultrasonic [4], eddy current [5]–[7], infrared thermography [8]–[10], optical [11]–[14], electrical [15], [16], and electrochemical testing [17]-[29] methods. All these methods have their own advantages, however, each of them also has its particular limitations. For example, a smooth or polished surface is required for ultrasonic and eddy current techniques; the electrical resistance (ER) sensor technique is more suitable for measuring uniform corrosion than localized corrosion; the infrared thermography technique has disadvantages such as a time-consuming process and the need for relatively expensive microwave equipment [8]; and the embedding of optical fibers runs the risk of decreased coating protectiveness through decreased adhesion or increased defect formation [30]. More importantly, these aforementioned methods are incapable of characterizing and monitoring the early degradation of organic coatings or being applied in field monitoring.

Electrochemical techniques have been successfully used in corrosion and coating studies over the past several decades. Among these techniques, electrochemical impedance spectroscopy (EIS) is the most popular for investigating the protective properties of coatings systems on metals. EIS is

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a nondestructive method that can be used to evaluate the health of organic coatings. The advantages of EIS are its precise, stable, and reliable performance and ability to provide complete and detailed impedance information regarding coatings. This information can be used to evaluate the health of coatings and accordingly schedule early maintenance. The disadvantages of EIS include its high cost; bulky hardware; and requirement of a low-disturbance environment, steady power supply, and other additional equipment; these disadvantages limit its usability for in situ measurements. As a result, many researchers have aimed to simplify the EIS experimental setup and accordingly proposed simplified EIS-based monitoring systems [18]–[29]. For example, Sebar et al. recently formulated a portable low-cost device for performing EIS measurements [29]. This instrument can measure the impedance in the frequency range from 0.01 Hz to 50 kHz. Further, the system uncertainty is less than 5% for an impedance of up to 50 k Ω . Unfortunately, they did not test the instrument's accuracy on coated samples; they only used a known capacitor, resistor, and metallic substrate. In addition, the measurement of a wide frequency range (0.1 Hz to 10k Hz) to obtain the full impedance spectra may limit the possibility of further reducing the measurement time. In 1999, Davis et al. proposed an EIS-based in situ corrosion sensor called a coating health monitor (CHM) to detect coating defects and to monitor the coating degradation in Army ground vehicles. This simplified impedance measuring system was designed to take three single-frequency electrochemical impedance measurements at 0.2, 0.5, and 0.9 Hz, respectively. The CHM results indicated that the EIS measurements at three frequencies agreed well with those obtained using a commercial Gamry potentiostat (Warminster, PA, USA) for a resistor-capacitor circuit and a coated specimen [18]. Although the CHM is a batterypowered, wireless microsensor system with EIS functionality, it may not be suitable for onsite monitoring because of its high cost and limited performance (e.g., CHM cannot measure a coating with an impedance above $5 \times 10^8 \ \Omega$ -cm²) [24]. Lee et al. proposed a prototype miniaturized impedance measurement instrument and demonstrated that it can provide a quick estimate of the protective capacity of coatings [24].

In our previous study, an Altera DE0-Nano fieldprogrammable gate array (FPGA) was employed for developing an innovative, compact, and portable second-generation coating impedance detector (CID 2.0) [25]. CID 2.0 consists of an FPGA and a custom printed circuit board (PCB) that mainly performs the voltage division used to calculate the impedance of coatings. Fig. 1 displays the measurement setup of CID 2.0. An oscilloscope and signal generator were successfully integrated in CID 2.0 by using the FPGA. CID 2.0 has dimensions of only $5 \times 9 \times 0.16$ cm³, making it considerably smaller than the first-generation CID and a conventional potentiostat system. Previous experimental results indicated that the FPGA-based CID 2.0 can reliably measure impedance values of 10^6 to $10^{10} \Omega$ -cm².



FIGURE 1. Measurement configuration of CID 2.0 [24].

Although CID 2.0 can detect variations in the coating impedance value in the range of 10^5 to $10^9 \Omega$ (as raw measurements without normalization by the subject area of 7.8 cm^2), its feasibility and accuracy for measuring a thick and high-performance coating having impedance exceeding $10^9 \Omega$ remain a serious concern. Fig. 2 indicates that although the average impedance value measured by CID 2.0 $(4.51 \times 10^9 \Omega)$ is close to that obtained using a commercial potentiostat (2.78 \times 10⁹ Ω), CID 2.0 measurements had a standard deviation of up to $3.01 \times 10^9 \Omega$, suggesting that these measurements could be unreliable when the raw impedance value of the examined samples exceeds $10^9 \Omega$. To reduce data uncertainty and increase the measurement reliability when the examined coatings have high impedance values, as is the case with the commercial paints applied within the offshore wind energy industry, this study proposes CID 3.0, a revised version of CID 2.0. By modifying the analog part of CID 2.0 and using oversampling, the overall noise is reduced to a lower level with CID 3.0. This feature is expected to allow CID 3.0 to achieve a much higher accuracy and working range compared with CID 2.0. More importantly, these notable system performance improvements achieved using the newly designed detector incur no additional cost and require no substantial change in size, suggesting that CID 3.0 would be more suitable than CID 2.0 for practical applications in the health monitoring of extremely robust coatings in the field. In this study, the impedance of commercial coated samples measured using CID 3.0 was recertified and compared with that of samples measured using the conventional potentiostat. Continuous monitoring experiments were also conducted to demonstrate the efficiency of CID 3.0 in monitoring the coating degradation.

II. SYSTEM DESIGN

The previous version of our instrument uses two unity-gain buffer circuits and two analog-to-digital converters (ADCs) to measure the input voltage signal and the impedancedivided voltage signal [Fig. 3(a)]. The goal was to reduce



FIGURE 2. Comparison of results obtained using CID 2.0 and conventional potentiostat for a high-performance commercial coating.

the need for calibration without using any external resistors other than the reference resistor. However, the ADCs have internal preamplifiers, and mismatches exist between the two ADCs. Consequently, calibration would still be required to improve measurement accuracy. In this study, an improved version with higher measurement accuracy was implemented. Because calibration is unavoidable if one wishes to achieve high accuracy, the measurement of the input signal is removed; thus, only one operational amplifier and one ADC are required, resulting in a 50% lower power consumption. Moreover, the gain can now be provided by the operational amplifier to improve the signal-to-noise ratio (SNR) of the signal going into the ADC. Therefore, the new circuit has lower power consumption with fewer components and better accuracy through the improved SNR and calibration compared with the previous implementation.

Fig. 3(b) presents the block diagram of CID 3.0. The FPGA generates a continuous 10-mV sinusoidal signal with a frequency of 0.5 Hz (V_1 in Fig. 3(b)). According to the Friis formula, when the overall gain of a cascade of stages remains the same, the overall noise figure is lower if the first stage's gain is higher. CID 2.0 used two unity buffers at the first stage of the system. In CID 3.0, these unity buffers are replaced with a noninverting amplifier to reduce the overall noise and simultaneously ensure v1 is unaffected by the loading effect.

To further reduce noise, the sampling rate, $f_{sampling}$, of the ADC was increased. When the sampling rate is higher, the spreading interval of noise increases. Therefore, the quantization noise is divided by the oversampling ratio (OSR). The relationship between OSR and SNR can be expressed as

$$OSR = \frac{f_{sampling}}{2 \times f_{signal}}$$
(1)

$$SNR(dB) = 1.76dB + 6.02N + 10log(OSR)$$
 (2)

where f_{signal} is 0.5 Hz in CID 3.0. Based on (2), by increasing $f_{sampling}$, signals are less affected by noise in CID 3.0 than in CID 2.0. According to Kirchoff's law, V_2 can be expressed as

$$V_2(0.5\text{Hz}) = V_s(0.5\text{Hz}) \times \text{G} \times \frac{Z_{dut}}{R_{ref} + Z_{dut}}$$
(3)



FIGURE 3. Block diagram of (a) CID 2.0 and (b) CID 3.0.

where V_2 (0.5 Hz) and V_S (0.5 Hz) are the Fourier transforms of $V_2(t)$ and $V_s(t)$, respectively, at 0.5 Hz, and G is the noninverting amplifier's gain, which can be calculated as

$$G = \frac{R1 + R2}{R1} \tag{4}$$

When the coated sample is not connected to the circuit, the circuit is open and Z_{dut} in (3) approaches infinity. Therefore, V_2 can be expressed as

$$V_2(0.5\text{Hz})|_{Z_{dut}\to\infty} = V_s(0.5\text{Hz}) \times G$$
(5)

Finally, the relationship between Z_{dut} and R_{ref} can be expressed as

$$\frac{Z_{dut}}{R_{ref} + Z_{dut}} = \frac{V_2(0.5\text{Hz})}{V_2(0.5\text{Hz})|_{Z_{dut} \to \infty}}$$
(6)

Therefore, Z_{dut} can be found by solving (6). Notably, R_1 and R_2 are not variables in (6). Therefore, their precise values need not be known to calculate Z_{dut} . The resistors R_1 and R_2 solely provide gain to the input signal and thus improve the noise performance.

III. EXPERIMENTAL PROCEDURE

Seven ideal resistors with resistances ranging from 10^4 to $10^{10} \Omega$ and various coating system samples with impedance values (raw measurement) exceeding $10^9 \Omega$, including aluminum substrates with a 75- μ m-thick epoxy coating system

and two 450- and 550- μ m-thick high-performance commercial coated steels, were prepared to verify the accuracy and performance range of the proposed device. For comparison, the impedance was also measured using a conventional potentiostat (Ref 600, Gamry Instruments, Pennsylvania, USA). A potentiostat with a standard three-electrode setup was used; it comprised the coated sample as the working electrode, a saturated calomel reference electrode, and a graphite rod as the counter electrode. CID 3.0 was used with a simplified two-electrode setup that is more suitable for installation in the field. The exposed surface area of the coated sample was 0.78 cm², and the working electrolyte was a 3.5 wt% NaCl solution. To evaluate the monitoring function, a knife was used to scratch the coated samples (length of the scratch: 5 mm) to create an artificial defect in both lab and field EIS monitoring tests. After the artificial defect was produced, the impedance value of the damaged sample was measured using a potentiostat and CID 3.0.

IV. RESULTS AND DATA ANALYSIS

A. HARDWARE IMPROVEMENT

To estimate the accuracy of the CID 3.0 measurements, the impedance values of seven ideal resistors with resistances of 10^4 to 10^{10} Ω were measured using CID 3.0 and a commercial volt-ohm-milliammeter (VOM). Table 1 presents a summary of the measurement results. The impedance values of the seven ideal resistors as obtained using CID 3.0 were 9.78 k Ω , 99.4 k Ω , 0.992 M Ω , 9.91 M Ω , 101.18 M Ω , 1.072 G Ω , and 9.07 G Ω , respectively; these are similar to the values measured by VOM. This finding indicates favorable accuracy because the error of the impedance values between CID 3.0 and VOM is within 3%.

Fig. 4 illustrates Bode magnitude plots of three coated samples as measured using a potentiostat; the green points in the three figures indicate the average impedance value of five raw measurements at 0.5 Hz as obtained using CID 3.0. In Fig. 4(a), the $|Z|_{0.5Hz}$ value of the 75- μ m-thick epoxy-coated Al substrate as measured using the potentiostat was $3.03 \times 10^8 \Omega$. The average of five $|Z|_{0.5Hz}$ values obtained using CID 3.0 for the same coating sample was $3.64 \times 10^8 \Omega$ with a standard deviation of 5.57 \times $10^6 \Omega$. Moreover, in Fig. 4(b), the $|Z|_{0.5Hz}$ value of the 450- μ m-thick commercial coated steel substrate as measured using the potentiostat was $0.98 \times 10^9 \Omega$. The average of five |Z|_{0.5Hz} values obtained using CID 3.0 for the same coating sample was $1.11 \times 10^9 \Omega$ with a standard deviation of $1.69 \times 10^7 \Omega$. Notably, for the 550- μ m-thick commercial coated steel substrate (Fig. 4(c)), the $|Z|_{0.5Hz}$ value of the coated sample as measured using the potentiostat was as high as $1.37 \times 10^{10} \Omega$, and the average of five $|Z|_{0.5Hz}$ values obtained using CID 3.0 for the same coating sample was $1.35 \times 10^{10} \Omega$ with a standard deviation of $2.02 \times 10^9 \Omega$. The measurement results obtained using CID 3.0 and the potentiostat for the three coated samples were in agreement with each other. Overall, the performance of CID 3.0 was comparable to that of CID 2.0. Moreover, the accuracy and

TABLE 1. Measurement data of commercial VOM and CID 3.0.

Resistor (Ω)	VOM	CID 3.0	Standard deviation	Error (%)
10k	10.01k	9.78k	3.3k	2.30
100k	99.7k	99.4k	1.15k	0.29
1M	0.990M	0.992M	12.4k	0.19
10M	9.87M	9.91M	47.6k	0.45
100M	101.20M	101.18M	0.23M	0.02
1G	1.075G	1.072G	4.75M	0.28
10G	9.10G	9.07G	0.43G	0.35

TABLE 2.	Measurement	data	of long	g-term	corrosion	monitoring
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Sample (Ω)	Potentiostat	CID 3.0
1 h	1.33×10^{10}	1.35×10^{10}
24 h	1.14×10^{10}	1.12×10^{10}
36 h	1.13×10^{10}	1.09×10^{10}
48h(with scratch)	8.55×10^{9}	7.74×10^{9}
60h(with scratch)	5.31×10 ⁹	4.56×10 ⁹

stability of CID 3.0 for estimating the impedance values of high-quality coatings were higher than those of CID 2.0.

B. MONITORING DEGRADATION OF PROTECTIVE COATINGS

After the successful improvement of the CID measurement accuracy for measuring the high-performance coating, the feasibility of using CID 3.0 to periodically detect the impedance values of coatings required reconfirmation to ensure coating degradation could be detected. Fig. 5 presents the Bode plots (solid and dashed colored line) measured using the potentiostat and single-frequency impedance value (colored points) measured using CID 3.0 after immersion at various durations. First, in the Bode plots obtained using potentiostat measurements, the epoxy coating exhibited almost an inclined line with high impedance value (exceeding $10^{10} \Omega$ -cm² at the lowest frequency of 10^{-2} Hz) at the beginning of the test, indicating its excellent barrier property. Subsequently, an artificial scratch was introduced on the coating to initiate the degradation of the protective coating 48 h after immersion; subsequently, the impedance value of the damaged coating was measured (dashed lines in Fig. 5). The substantial decrease in low-frequency impedance values for a damaged coating sample after 60 h of immersion clearly indicated that the loss of barrier properties occurred because of the penetration of water and electrolyte through the pores and defects of the coating [31]. Similar to the $|Z|_{0.5Hz}$ value measured using the potentiostat, the $|Z|_{0.5Hz}$ value abruptly decreased immediately in the measurement result obtained using CID 3.0, suggesting that CID 3.0 can sensitively measure changes in the coating impedance value as well. Table 2 summarizes the measurement data of $|Z|_{0.5Hz}$ values obtained using CID 3.0 and the potentiostat.



FIGURE 4. Comparison of results obtained using CID 3.0 and conventional potentiostat for (a) 75- μ m-thick epoxy-coated Al substrate, (b) 450- μ m-thick commercial coated steel substrate, and (c) 550- μ m-thick commercial coated steel substrate.

 TABLE 3. Measurement data of long-term corrosion monitoring in field environment.

Sample (Ω)	Potentiostat	CID 3.0
1 h	1.00×10^{10}	9.62×10 ¹⁰
24 h	1.02×10^{10}	9.73×10 ¹⁰
48 h	8.83×10 ¹⁰	8.26×10 ¹⁰
49h(with scratch)	5.82×10 ⁹	5.62×10 ⁹
72h(with scratch)	4.83×10^{9}	4.83×10 ⁹

C. FIELD TESTING OF PROTECTIVE COATINGS

In addition to the corrosion monitoring results obtained under laboratory conditions, the feasibility of using CID 3.0 to



FIGURE 5. Results of long-term corrosion monitoring in lab environment.



environment.

detect changes in the impedance of coatings under field conditions was also evaluated in this study. The coated samples were prepared and positioned with a tilt angle of 30° under outdoor atmospheric conditions, following which they underwent dry and rainy periods during these evaluations. Fig. 6 presents the Bode plots measured using the potentiostat (solid and dashed colored line) and singlefrequency impedance value (colored points) measured using CID 3.0 after exposure at various durations. The consistent Bode plots and stable impedance results obtained using potentiostat measurements and CID 3.0 both indicated that the coatings retained their qualities as an excellent barrier after the first 48 h of exposure. After an artificial scratch was made on the coating 48 h after exposure, the low-frequency impedance values of the damaged coating sample after 1 (the 49th hour) and 24 (the 72th hour) h of exposure substantially decreased, indicating that the coating loses its protection barrier as a result of the scratch. Similar to the $|Z|_{0.5Hz}$ value measured using the potentiostat, the $|Z|_{0.5Hz}$ value abruptly decreased in the measurement result obtained using CID 3.0. This experiment demonstrated the feasibility of CID 3.0 for

use in the field. Table 3 summarizes the measurement data of $|Z|_{0.5Hz}$ values obtained using CID 3.0 and the potentiostat in the field environment. Overall, although CID 3.0 could not provide a detailed performance assessment of coatings and provide kinetic information regarding the corrosion process, it is highly useful to engineers in determining whether coating maintenance should be scheduled through the monitoring of changes in the coating impedance values.

V. CONCLUSION

A practical and reliable CID 3.0, an improved version of CID 2.0, was proposed and characterized in this study for the rapid detection of coating degradation. The consistency between the impedance values measured using CID 3.0 and the commercial VOM for a given resistor in the range of 10^3 to $10^9 \Omega$ demonstrated the reliability of CID 3.0. More importantly, compared with CID 2.0, CID 3.0 exhibited a higher estimated accuracy when raw measurements were conducted on the high-performance coated sample with impedance values exceeding $10^9 \Omega$, thus extending the applicability of CID 3.0 to diverse commercial coating systems. The corrosion monitoring measurement results also demonstrate that CID 3.0 can detect changes in the impedance value of coatings when they degrade in a corrosive environment. A CID with wireless communication function (CID 4.0) will be studied in the future.

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