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# **Sheet Resistance Reduction of MoS<sup>2</sup> Film Using Sputtering and Chlorine Plasma Treatment Followed by Sulfur Vapor Annealing**

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**ABSTRACT** Sheet resistance  $(R_{sheet})$  reduction of a-few-layered molybdenum disulfide  $(MoS_2)$  film using sputtering is investigated in this study. To enhance the carrier density, chlorine  $(Cl<sub>2</sub>)$  gas excited by inductively coupled plasma is introduced as a substitute for sulfur. To electrically activate the Cl dopants and simultaneously prevent out-diffusion of sulfur, a furnace annealing was performed in sulfur-vapor ambient. Consequently, the  $R_{sheet}$  in the MoS<sub>2</sub> film with the Cl<sub>2</sub> plasma treatment remarkably reduced by one order lower than that without one, because of the activation of Cl dopants in the  $MoS<sub>2</sub>$  film.

**INDEX TERMS** Activation annealing, chlorine plasma, molybdenum disulfide (MoS<sub>2</sub>), radio-frequency magnetron sputtering, transition metal dichalcogenide (TMDC).

#### **I. INTRODUCTION**

 $MoS<sub>2</sub>$  film which is one of the transiton metal dichalcogenides (TMDCs) has attracted great attentions, because of its excellent electrical and physical properties such as its high mobility even at atomically thin thickness, adequate band-gap, flexibility and transparency [\[1\]](#page-5-0)–[\[4\]](#page-5-1) for advanced LSIs, energy harvesters, displays and sensors [\[5\]](#page-5-2)–[\[10\]](#page-6-0). To obtain an atomically thin  $MoS<sub>2</sub>$  film with large area for such applications, a chemical vapor deposition (CVD) method is considered. However, to synthesize large  $MoS<sub>2</sub>$  film on substrate, a special treatment with alkali metal is used [\[11\]](#page-6-1), [\[12\]](#page-6-2), that influences in the Fermi-level pinning near the conduction band minimum [\[13\]](#page-6-3). As a method to avoid the unexpected difficulties and achieve large films, a sputtering method under ultra high vacuum (UHV) has been proposed as a physical vapor deposition (PVD) [\[14\]](#page-6-4), [\[15\]](#page-6-5). For the Seebeck device in thermoelectric generator as an energy harvester, high efficiency of energy conversion was achieved in a sputtered  $MoS<sub>2</sub>$  film by low thermal conductivity [\[16\]](#page-6-6). However, sulfur

atoms are easily out-diffused from the  $MoS<sub>2</sub>$  film during sputtering process, which cause high carrier density. We have found that sulfur vapor annealing (SVA) compensates for S defects resulting in a carrier density reduction down to 1.8  $\times$  10<sup>16</sup> cm<sup>-3</sup> and also a mobility enhancement [\[17\]](#page-6-7), [\[18\]](#page-6-8). Furthermore, normally-off nMISFETs with an appropriate threshold voltage were performed, because of the low carrier density in  $MoS<sub>2</sub>$  film formed by sputtering and treated by SVA and an appropriate work function of metal gate [\[19\]](#page-6-9).

To intentionally enhance the carrier density in low carrierdensity  $MoS<sub>2</sub>$  film for the Seebeck device, intrinsic and extrinsic carrier generations are expected. Although the intrinsic technique such as vacancies and interstitials of the consisting materials generates several energy levels in the band gap [\[20\]](#page-6-10)–[\[22\]](#page-6-11), the extrinsic technique is favorable rather than intrinsic one to significantly enhance the carrier density with high controllability [\[23\]](#page-6-12), for which electrostatic doping [\[24\]](#page-6-13), molecular adsorption [\[25\]](#page-6-14)–[\[27\]](#page-6-15) and substitutional doping [\[28\]](#page-6-16)–[\[33\]](#page-6-17) can be considered. Although electrostatic

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doping has high carrier density, a device structure is complex. And molecular adsorption is unstable on thermodynamics. Therefore, substitutional doping technology is selected in this study, because of its simple structure and stability. Since high carrier mobility is required to enhance the power factor (PF) of the Seebeck device, an n-type  $MoS<sub>2</sub>$  film is selected rather than p-type one [\[34\]](#page-6-18), [\[35\]](#page-6-19). As candidates for the ntype dopant, Group 7 and 17 elements were examined as substitution for molybdenum (Mo) and S, respectively. In terms of ionization and formation energies, chlorine (Cl) substituting for S was selected as the n-type dopant in the  $MoS<sub>2</sub>$  film [\[36\]](#page-6-20), [\[37\]](#page-6-21). For the fabrication method, ion implantation and plasma exposure in dopant ambient have been investigated [\[28\]](#page-6-16)–[\[31\]](#page-6-22). However, energetic dopants in these methods can generate S vacancies and an activation of dopants in the  $MoS<sub>2</sub>$  film is required [\[31\]](#page-6-22).

In this study, we investigates the combination of Cl2-plasma treatment and SVA for sheet resistance reduction in  $MoS<sub>2</sub>$  films formed by sputtering.

#### **II. EXPERIMENTAL METHODS**

To examine the sheet resistance in the  $MoS<sub>2</sub>$  film, a circular transmission line model (CTLM) pattern was used, as shown in Fig. [1](#page-1-0) [\[38\]](#page-6-23)–[\[40\]](#page-6-24). A base material of silicon dioxide  $(SiO<sub>2</sub>)$ on a silicon substrate was cleaned in a wet process using a piranha solution. Titanium-nitride (TiN) electrodes of 40 nm thickness were formed by sputtering and wet etching with  $H<sub>2</sub>O<sub>2</sub>$ . The MoS<sub>2</sub> films were formed using an ultra-high vacuum (UHV) radio frequency (RF) magnetron sputtering tool with  $MoS<sub>2</sub>$  target of 99.99% purity, at a substrate temperature of 300◦C, under argon (Ar) pressure of 0.55 Pa, an Ar flow rate of 7 sccm, an RF power of 40 W and a distance of 150 mm between the  $MoS<sub>2</sub>$  target and the substrate. We note that the sputter-deposited  $MoS<sub>2</sub>$  film has layered structure in parallel to the substrate because of high temperature at 300◦C. The detail of the film was investigated in our previous work [\[18\]](#page-6-8). The  $MoS<sub>2</sub>$  films were exposed to a Cl<sub>2</sub> plasma generated using an inductive coupling plasma reactive ion etching (ICP-RIE) tool (SAMCO RIE-101iPH), under  $Cl_2$ ) pressures from 0.4 to 3.2 Pa, with a  $Cl_2$  flow rate of 7 sccm, ICP powers from 5 to 20 W, for 20 to 160 s. Herein, the bias power is set to 0 W resulting in a floating potential at the sample. The  $MoS<sub>2</sub>$  films were annealed by SVA at the 700 °C in a sulfur ambient under 100 Pa for 40 min to activate the dopants and compensate for the sulfur vacancies, simultaneously. The thickness of the  $M_0S_2$ film was approximately 2.0 nm measured using the X-ray reflection (XRR) method. Sheet resistances in the  $MoS<sub>2</sub>$  film were extracted from the CTLM measurement with spacing from 5 to 30  $\mu$ m and r = 90  $\mu$ m.

Raman spectroscopy was performed at 532 nm wavelength and pseudo-Voigt function was used to fit the Raman spectrum of the  $MoS<sub>2</sub>$  film. An atomic force microscope (AFM) was performed in dynamic force microscope (DFM) mode. The changes in the thickness of the  $MoS<sub>2</sub>$  layer with and



<span id="page-1-0"></span>**FIGURE 1. Schematic illustrations of the CTLM pattern with TiN contacts and MoS<sub>2</sub> film. (a) Cross-sectional and (b) top views with**  $r = 90 \mu m$  **and several contact distances** *d* **from 5 to 30 µm.**

without the  $Cl<sub>2</sub>$  plasma treatment were observed using transmission electron microscopy (TEM) with an accelerating voltage of 200 keV. An X-ray photoelectron spectroscopy (XPS) using an Al  $K_{\alpha}$  X-ray source was also performed with spectral fitting using the pseudo-Voigt function. The composition ratio of the  $MoS<sub>2</sub>$  film is expressed as follows:

$$
C_i = \frac{A_i/\text{RSF}_i}{\sum_j A_j \text{RSF}_j}
$$
 (1)

where  $C_i$ ,  $A_i$ , and  $RSF_i$  are the composition ratio, peak area, and relative sensitivity factor of the atom i, respectively [\[17\]](#page-6-7). Depth profiling of time-of-flight secondary ion mass spectroscopy (TOF-SIMS) were performed using bismuth and cesium ions as the primary and sputtering ion sources, respectively.

#### **III. RESULTS AND DISCUSSION**

To evaluate the sheet resistance reduction in the  $MoS<sub>2</sub>$  film, typical results from CTLM with and without  $Cl<sub>2</sub>$  plasma treatment are shown in Fig. [2,](#page-2-0) as a function of the contact distance at 5 W under 0.4 Pa for 40 s. The total resistance depending on contact distance is shown in following Equation [\(2\)](#page-1-1),

<span id="page-1-1"></span>
$$
R_{total} = \frac{R_{sheet}}{2\pi} \left[ \frac{L_t}{r} + \frac{L_t}{r+d} + \ln\left(1 + \frac{d}{r}\right) \right]
$$
 (2)

where  $R_{sheet}$ ,  $L_t$ ,  $r$ , and  $d$  are the sheet resistance, transfer length, inner radius, and contact distance, respectively. The  $R_{sheet}$  of 8.50 M $\Omega$ /sq. with the Cl<sub>2</sub> plasma treatment is



**FIGURE 2.** Resistance dependence on contact distance in MoS<sub>2</sub> films with **and without Cl2 plasma treatment at 5 W under 0.4 Pa for 40 s, followed by the SVA.**

<span id="page-2-0"></span>

<span id="page-2-1"></span>**FIGURE 3.** Sheet resistance of MoS<sub>2</sub> film as a function of Cl<sub>2</sub> plasma **exposure time. The duration time was varied at 5 W under 0.4 Pa. SVA was applied at 700◦C for 40 min.**

approximately ten times less than that of 105 M $\Omega$ /sq. without the treatment. Furthermore, the contact resistance normalized by the channel width decreases down to 1.07 k $\Omega$ .cm. Both sheet and contact resistances are significantly reduced by the  $Cl<sub>2</sub>$  plasma treatment.

In order to further reduce the resistance of the  $MoS<sub>2</sub>$ film, a sheet resistance dependence on plasma exposure time is shown in Fig. [3.](#page-2-1) *Rsheet* decreases and increases with an increase in plasma exposure time up to and beyond 40 s, respectively. It is speculated that the resistance reduction is because of an increase in the quantity of Cl dopants.

To discuss the cause of increase in the resistance, cross-sectional TEM images were examined, as shown in Figs. [4\(](#page-3-0)a), (b), and (c) for each exposure time of the  $Cl<sub>2</sub>$ plasma treatment. The thickness with  $Cl<sub>2</sub>$  plasma treatment for 40 s approximately equals to that without treatment. However, the thickness of the  $MoS<sub>2</sub>$  film with exposure time of 160 s is thinner than that of 40 s maintaining the layered

structure and the uniformity of the film thickness, as shown in Fig. [4\(](#page-3-0)c). This is because due to etching influences.

To explain the sheet resistance characteristics consisting of both doping and etching at the same time, we assume the following conditions. Initially, the conductance  $(G_n)$  of the *n*-th layer from the top layer of the  $MoS<sub>2</sub>$  film as a function of the exposure time  $t$  of the  $Cl_2$  plasma is expressed as

$$
G_n(t) = \frac{1}{R_n(t)} = A^{n-1}B(t - t_{inculation}) + G_{initial}, \qquad (3)
$$

where *A*, *B*, *tincubation*, and *Ginitial* are ratio of conductance increase on the number of layer, conductance increase over time, incubation time to stabilize the  $Cl<sub>2</sub>$  plasma and an incremental conductance at 0 s, respectively. If *t* was less than  $t_{\text{incubation}}$ ,  $G_n(t)$  is equal to  $G_{\text{initial}}$ . Therefore, the sheet resistance  $R_{sheet}(t)$  is given by

$$
R_{sheet}(t) = \frac{1}{\sum_n G_n(t)}.
$$
\n(4)

Furthermore, we assume that the top layer of the  $MoS<sub>2</sub>$ film is etched in the order with top-S, middle-Mo and bottom-S by the  $Cl_2$  plasma with a low ICP power [\[41\]](#page-6-25), as shown in Fig. [4\(](#page-3-0)c). Moreover, if the middle-Mo remained after etching, the S-Mo-S structure in the layer is going to be recovered by the SVA [\[17\]](#page-6-7). Under these speculations, if the middle-Mo in the top layer was not etched, the sheet resistance decreases with an increase in exposure time, because of the doping effect of less than 40 s, as shown in Fig. [3.](#page-2-1) In contrast, if the middle-Mo was etched, it is difficult to reduce the sheet resistance any more. Therefore these phenomena are expected in a cyclic manner, which is consistent with experimental results, as shown with dotted lines in Fig. [5](#page-3-1) Eventually, the minimum *Rsheet* at an exposure time of 40 s is approximately ten times less than that without treatment.

To confirm improvements in crystallinity of low resistance  $MoS<sub>2</sub>$  film, the Raman spectra with and without plasma treatment are shown in Fig. [6.](#page-3-2) The peaks of  $E_{2g}^1$  (in-plane vibration mode of Mo and S) and  $A_{1g}$  (out-of-plane mode vibration of S) were observed even after  $Cl<sub>2</sub>$  plasma treatment. Fig. [7](#page-3-3) shows the full width at half maximum (FWHM) values of  $E_{2g}^1$  and  $A_{1g}$  with and without the Cl<sub>2</sub> plasma treatment followed by the SVA. The FWHM values increase after the plasma treatment. this is due to the damaged crystallinity of  $MoS<sub>2</sub>$  film by the Cl<sub>2</sub> plasma treatment. On the other hand, although the FWHM values of the film significantly decrease after the SVA, the FWHM values with the  $Cl<sub>2</sub>$  plasma treatment are still greater than those without treatment, because of disorder of phonon can be considered due to Cl dopants.

Root mean square (RMS) values of the roughness obtained by the AFM, as shown in Fig. [8,](#page-3-4) increase after the plasma treatment and decrease by the SVA, whose trend is consistent with the changes in crystallinity discussed on the Raman spectra in Fig. [7.](#page-3-3)

To discuss the dopant characteristics, a resistivity depending on temperature is modeled using the thermally activated



**FIGURE 4. Cross-sectional TEM images of MoS2 film depending on Cl2 plasma exposure times of (a) no exposure, (b) 40 s, and (c) 160 s, each followed by SVA.**

16

 $FWHM$   $\text{cm}^{-1}$ 

<span id="page-3-0"></span>

<span id="page-3-1"></span>**FIGURE 5. Comparison of the model and experimental results on sheet resistance of the MoS<sub>2</sub> film as a function of the Cl<sub>2</sub> plasma exposure time.** The dotted line shows the calculated sheet resistance in the MoS<sub>2</sub> film from the model, and the value at 0 s corresponds to that without the Cl<sub>2</sub> **plasma treatment.**



<span id="page-3-2"></span>**FIGURE 6. Raman spectra of MoS<sub>2</sub> films with and without Cl<sub>2</sub> plasma treatment at 5 W under 0.4 Pa for 40 s followed by SVA.**

transport equation [\[42\]](#page-6-26):

$$
\rho = \frac{1}{\sigma} = \frac{1}{qn(T)\mu(T)} = \frac{1}{\sigma_0(T)} \exp\left(\frac{E_a}{k_B T}\right) \tag{5}
$$

 $E_{2g}^{\qquad 1}$ 15 13  $12$  $1^{\circ}$  $10$ 9 g w/ SVA as sputter w/ Cl<sub>2</sub> plasma

FIGURE 7. FWHM values in  $E^1_{2g}$  and  $A^1_g$  of the Raman spectra for MoS<sub>2</sub> **films along each process. The dotted lines connect the FWHM values** without the CI<sub>2</sub> plasma treatment.

<span id="page-3-3"></span>

<span id="page-3-4"></span>**FIGURE 8. RMS values in the AFM of MoS<sub>2</sub> film along process.** 

where  $E_a$ ,  $k_B$ , and  $\sigma_0$  are the activation energy, the Boltzmann constant, and pre-factor depending on the temperature, respectively. It has been reported that  $\sigma_0$  is insensitive at the temperature above 170 K [\[43\]](#page-6-27), [\[44\]](#page-6-28). Therefore, the Arrhenius plot of resistivity in the  $MoS<sub>2</sub>$  film was determined using Van der Pauw measurement with 1 cm  $\times$  1 cm MoS<sub>2</sub>/SiO<sub>2</sub>/Si samples patterned with TiN bottom contacts, as shown in Fig. [9.](#page-4-0) The resistivity values of the  $MoS<sub>2</sub>$  film below 170 K



**FIGURE 9.** Arrhenius plots of resistivity in MoS<sub>2</sub> film with and without Cl<sub>2</sub> **plasma treatment at 5 W under 0.4 Pa for 40 s, followed by SVA.**

<span id="page-4-0"></span>

<span id="page-4-1"></span>**FIGURE 10. XPS spectra of (a) Mo 3***d* **and (b) S 2***p* **in the MoS<sub>2</sub> film with and without Cl2 plasma treatment at 5 W under 0.4 Pa for 40 s, followed by SVA.**

and 190 K with and without the  $Cl<sub>2</sub>$  plasma treatment, respectively, were ignored due to the Schottky  $I - V$  characteristics. The activation energies with and without the  $Cl<sub>2</sub>$ 



FIGURE 11. S/Mo composition ratio of the MoS<sub>2</sub> film determined from the **Mo-S and S-Mo peak areas in the XPS spectra along each process.**

<span id="page-4-2"></span>

<span id="page-4-3"></span>**FIGURE 12. XPS spectra of Cl 2***p* **in the MoS2 films along each process.**

plasma treatment were 117 and 154 meV, respectively. To validate the polarity of the dopants, XPS analyses of the  $MoS<sub>2</sub>$  film surface with and without the  $Cl<sub>2</sub>$  plasma treatment were performed. From Figs. [10\(](#page-4-1)a) and (b), the binding energy peaks of Mo  $3d_{5/2}$  and S  $2p_{3/2}$  shift toward the positive direction. These explain that positive-increases in a position of binding energy peaks correspond to the Fermi level shift approaching to the conduction band minimum.

To discuss the cause of the Fermi level shift to the conduction band minimum, there are two main candidates of S vacancy and Cl doping. A S/Mo composition ratio in the  $MoS<sub>2</sub>$  film with the  $Cl<sub>2</sub>$  plasma treatment calculated from the Mo-S and S-Mo peak areas in the XPS, presented in Fig. [11,](#page-4-2) was approximately two. This means that the density of S vacancies in the film was significantly reduced by the SVA. In addition, the ratio slightly decreases with an increase in the duration time of the  $Cl<sub>2</sub>$  plasma treatment and a S/Mo ratio of 1.56 with the plasma treatment for 40 s. If the ratio in each layer of 4-layer sputter-deposited  $MoS<sub>2</sub>$  film had the same value of 1.66 however only the top-S in the top layer of the film was removed by the plasma, the ratio would be

<span id="page-5-4"></span>

Formation	Thickness [nm]	Contact	Dopant	Type	Hall effect mobility $\rm [cm^2 V^{-1} s^{-1}]$	Carrier density $\rm [cm^{-3}]$	$R_{sheet}$ $\lceil \Omega/\text{sq.}\rceil$	Calculated $\rho$ $\lceil \Omega$ cm	$\pi_c$ $[k\Omega/\mu m]$	Ref.
Sputtering	2.4	TiN		$\it n$	$\sim$		$8.50 \times 10^{6}$	6.0	$\times 10^4$ L.O7	This work
<b>CVD</b>		Ni /Au/Ni	Nb		8.5	$3.1 \times 10^{20}$	$1.8\times10^3$	$2.4 \times 10^{-3}$	0.6	[45]
<b>CVD</b>	61	Au/Ti	Nb		14	$3.0 \times 10^{19}$	$2.5 \times 10^{3}$	$1.5 \times 10^{-2}$	$\blacksquare$	$[46]$
<b>CVD</b>	$.9 \times 10^5$	Au/Ti	Nb		7.5	$4.3 \times 10^{19}$		$1.9 \times 10^{-2}$	4.98	$[47]$
<b>CVD</b>		Аg			0.53	$1.0 \times 10^{19}$		1.2	$\blacksquare$	$[48]$
Exfoliation			Benzyl viologen	$\boldsymbol{n}$	2.39	$2.97 \times 10^{14}$	$1.0 \times 10^6$	$4.4 \times 10^{2}$		[49]

TABLE 1. Comparisons of electric characteristics in MoS<sub>2</sub> film with various dopants and contact metals, whose mobility and carrier density are **simultaneously obtained by the hall effect measurement.**



<span id="page-5-3"></span>**FIGURE 13. TOF-SIMS depth profiles from MoS<sub>2</sub> film in special structure, consisting of Al2O3 film on MoS2/SiO2/Si with Cl2 plasma treatment at 5 W under 0.4 Pa for 40 s, followed by the SVA.**

reduced down to 1.45. Therefore the etching effect discussed on the sheet resistance reduction is supposed. Fig. [12](#page-4-3) shows the Cl  $2p$  peaks of Cl<sub>2</sub> plasma treated MoS<sub>2</sub> films together with each process procedure. Although the Cl peaks were detected after the  $Cl_2$  plasma treatment, it is difficult to find them after the  $Cl<sub>2</sub>$  plasma treatment and the SVA, due to the detection limit of the XPS evaluation method. Therefore, the TOF-SIMS was performed to determine the existence of chlorine in the  $MoS<sub>2</sub>$  film with  $Cl<sub>2</sub>$  plasma treatment after SVA, for which an  $Al_2O_3$  film was deposited on the  $MoS_2$ film after the SVA. From the depth profiles in Fig. [13,](#page-5-3) a Cl ion peak is clearly observed in the  $MoS<sub>2</sub>$  film located near the surface without any adsorption on the surface. Moreover, considering with the electrical properties, Cl atoms in the  $MoS<sub>2</sub>$  film are successfully activated as n-type dopant during the SVA.

As a benchmark, Table [1](#page-5-4) shows comparison of electric characteristics between our result and  $MoS<sub>2</sub>$  films applied various doping techniques. Resistivities were calculated from a carrier density and the Hall mobility. Low resistivities in p-type are attributed to degenerative doping over  $10^{19}$  cm<sup>3</sup>. In contrast, non-degenerative doping is considered for our technique because the sheet resistance in the doped film is ten times smaller than that with only the SVA, whose carrier density is considered as around  $10^{16}$  cm<sup>3</sup> [\[17\]](#page-6-7). On the other hand, low contact resistances for p-type film have been also reported. Although the obtained contact resistance in our n-type work is higher than those due to a wide Schottly barrier width by the relatively low carrier density, the doping technology using combination of  $Cl<sub>2</sub>$  plasma and SVA in this article can enable us a good controllability in order of  $10^{17}$ to  $10^{18}$  cm<sup>-3</sup> of carrier density.

#### **IV. CONCLUSION**

Sheet resistance reduction in the  $MoS<sub>2</sub>$  film by a combination of Cl<sup>2</sup> plasma and SVA was investigated in the atomic thin film region. The sheet resistance of 8.50 M $\Omega$ /sq. with Cl<sub>2</sub> plasma treatment, which is approximately ten times smaller than that without treatment, was observed. The  $MoS<sub>2</sub>$  film was successfully doped in n-type with Cl dopants. This technique is promising to control the carrier density in the MoS<sup>2</sup> film for various applications, such as MISFETs and thermoelectric devices.

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